

# Quantum chemistry Study Questions Guide

1. What is quantum mechanics used to describe?
  - A) Large objects
  - B) Macroscopic objects
  - C) Microscopic objects
  - D) Astronomical objects
2. What is the Schrödinger equation used for?
  - A) Describing the behavior of macroscopic objects
  - B) Describing the behavior of microscopic objects
  - C) Describing the behavior of celestial bodies
  - D) Describing the behavior of subatomic particles
3. What is the wave-particle duality of quantum mechanics?
  - A) Particles can act as waves and vice versa
  - B) Particles can only act as waves
  - C) Waves can only act as particles
  - D) There is no wave-particle duality
4. What is the Heisenberg uncertainty principle?
  - A) The more precisely the position of a particle is known, the less precisely its momentum can be known
  - B) The more precisely the momentum of a particle is known, the less precisely its position can be known
  - C) The position and momentum of a particle can be known precisely at the same time
  - D) The position and momentum of a particle are unrelated

# Answer Key Study Guide Quantum Theory

**Douglas Brandt**

## Answer Key Study Guide Quantum Theory:

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