

Density Worksheet
Physical Science
 $D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - A. mass = 453 g; volume = 225 cm³
 - B. mass = 5.0 g; volume = 10.0 cm³
 - C. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Answers To Aventa Physical Science

N Noddings



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Physical Science National Learning Corporation, 2018 The DSST Physical Science Passbook R prepares candidates for the DSST exam which enables schools to award credit for knowledge acquired outside the normal classroom environment It provides a series of informational texts as well as hundreds of questions and answers in the areas that will likely be covered on your upcoming exam including but not limited to physics electricity and magnetism Glossyr chemical reactions atomic structure and more

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Mark A. Carle, 1991 *Physical Science Made Easy* Theodore G. Phillips, William Haskel Colburn, 1950 **Introduction to Physical Science** Alfred Payson Gage, 1903 **Physical World an Introduction to Physical Science** Donovan

Hadaway, Bob Jones University Press, 2005-06 The Physical World An Introduction to Physical Science for Christian Schools was written for you the curious student We have filled this textbook with answers to puzzling questions about why things happen and how things work But this text should do more than simply answer your questions It is intended to stimulate new questions that will cause you to expand your knowledge You will be introduced to realms that you have never before explored This book will take you inside an atom let you see what happens in solids liquids and gases help you to discover the forces that make things move or keep things from moving and show you forms of matter and energy that scientists are just beginning to understand Introduction **An Approach to Physical Science** Physical Science for Nonscience Students

Project, 1969 *Senior Physical Science*, 1969 *Experiences in Physical Science* Laidlaw Brothers, 1983 **Introduction to Physical Science** Alfred Payson Gage, 1889 *A First Course in Physical Science* Robert P. Bauman, 1987 This text provides an understanding of the foundations and structure of physical science by emphasizing science as a search for truth rather than an accumulation of facts It develops the subject through concrete examples such as inclined planes and levers speeds and displacements progressing to consideration of forces and the concept of inertia and the idea of energy Similarly a study of observable chemical reactions advances to the ways in which atoms combine separate and displace one another and how observable masses of reactants and products illustrate how the atoms are combining The fundamental ideas are applied to astronomy optics geology music and the chemical compounds of life processes The wide variety of end of chapter problems and multiple choice questions reinforce comprehension of each topic Physical Science Jerry Schad, 1996 Schad

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