

Chemical Reactions:

1. The following reaction $\text{PbO}_2 + \text{H}_2\text{O}_2 \rightarrow \text{PbO}_2\text{O}_3 + \text{H}_2\text{O}$ is not true. Which is not true?

A) reaction with 2.0 mol/l H_2O_2
 B) ground reaction with 3.0 mol/l H_2O_2
 C) reaction with 1.0 mol/l H_2O_2
 D) ground reaction with 2.0 mol/l H_2O_2
 E) reaction with 4.0 mol/l H_2O_2

2. A 1.0 mol/l example of the NO_x reaction is oxygenated according to the equation: $2\text{NO}_2\text{O}_2(\text{g}) \rightarrow 2\text{NO}_2\text{O}(\text{g}) + 2\text{O}_2(\text{g})$. Below, which reaction of O_2 with NO_2 will result in the same amount of NO_2 gas?

A) 0.1 mol
 B) 1.0 mol
 C) 1.2 mol
 D) 0.7 mol
 E) 2.0 mol

3. The reaction of NO_2 is step characterized by the equation: $4\text{NO}_2 + \text{O}_2 \rightleftharpoons 2\text{NO}_2\text{O}_2$. If reaction 2.0 mol/l NO_2 and 0.5 mol/l O_2 is carried out, then which reaction will affect the reaction rate most strongly?

A) 0.5 mol/l
 B) 0.7 mol/l
 C) 1.0 mol/l
 D) 1.5 mol/l
 E) 2.0 mol/l

4. Calculated the following reaction, v from 2000 mol/l NO_2 and 2000 mol/l O_2 : $2\text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g}) + \text{NO}_2\text{O}_2(\text{g})$

It is PbO_2 g and 20 mol/l H_2O_2 added to remove. In the following reaction, the production of 4.0 mol/l of H_2O_2 , which is the following reaction?

A) O_2
 B) NO_2
 C) NO_2
 D) NO_2
 E) C

5. Under the following conditions, $\text{NO}_2 + \text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g}) + \text{NO}_2\text{O}_2(\text{g})$ is not true. Which is not true?

A) reaction with 2.0 mol/l O_2 and 2.0 mol/l NO_2
 B) ground reaction with 3.0 mol/l O_2 and 2.0 mol/l NO_2
 C) reaction with 1.0 mol/l O_2 and 2.0 mol/l NO_2
 D) ground reaction with 2.0 mol/l O_2 and 2.0 mol/l NO_2
 E) reaction with 4.0 mol/l O_2 and 2.0 mol/l NO_2

6. The following reaction $2\text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g})$ is not true. Which reaction of O_2 with NO_2 will result in the same amount of NO_2 gas?

A) 1.0 mol/l
 B) 2.0 mol/l
 C) 0.5 mol/l
 D) 0.1 mol/l
 E) 0.2 mol/l

7. Under the following conditions, $\text{NO}_2 + \text{O}_2 \rightleftharpoons \text{NO}_2\text{O}_2$ is not true. Which reaction of O_2 with NO_2 is not true?

A) 0.5 mol/l O_2 and 0.5 mol/l NO_2
 B) 0.5 mol/l O_2
 C) 0.5 mol/l O_2
 D) 0.5 mol/l O_2
 E) 0.5 mol/l O_2

8. Under reaction conditions and quantities of reaction, the following reaction of NO_2 and O_2 is not true. Which reaction of O_2 with NO_2 is not true?



A) 1.0 mol/l O_2 and 2.0 mol/l NO_2
 B) 1.0 mol/l O_2 and 2.0 mol/l NO_2
 C) 0.5 mol/l O_2 and 2.0 mol/l NO_2
 D) 0.5 mol/l O_2 and 2.0 mol/l NO_2
 E) 0.5 mol/l O_2 and 2.0 mol/l NO_2

9. Calculate the mass of carbon dioxide produced from 2000 mol/l NO_2 and 2000 mol/l O_2 under the following reaction.



A) 7.0 g CO_2
 B) 1.0 g CO_2
 C) 0.5 g CO_2
 D) 1.2 g CO_2
 E) 1.800 g CO_2

10. Under the following conditions of oxygen, prepared to react with 1.0 g of sulfur in the following reaction.



A) 0.5 mol/l O_2 and 2.0 mol/l SO_2
 B) 0.5 mol/l O_2 and 2.0 mol/l SO_2
 C) 1.71 $\times 10^{-2}$ mol/l O_2 and 2.0 mol/l SO_2
 D) 0.402 $\times 10^{-2}$ mol/l O_2 and 2.0 mol/l SO_2
 E) 0.123 $\times 10^{-2}$ mol/l O_2 and 2.0 mol/l SO_2

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