



American History Political Cartoons Answer

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following mechanism
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ fast
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ slow
a write the overall balanced chemical equation
 $2 \text{O}_3 \rightarrow 3 \text{O}_2$ b identify any intermediates within the mechanism
c what is the order with respect

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 $k_1 \text{A} + 2 \text{B} \rightarrow \text{C}$
consider the following mechanism
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ fast
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ slow
a write the overall balanced chemical equation
 $2 \text{O}_3 \rightarrow 3 \text{O}_2$ b identify any intermediates within the mechanism
c

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 $\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
g the reaction between mg s and hcl a q is represented by the equation above in a kinetics experiment a 0.080 g sample of mg s ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M HCl a q at 25 °C

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