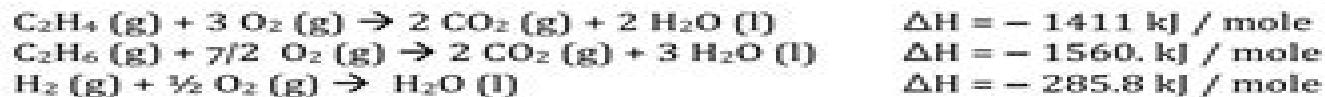


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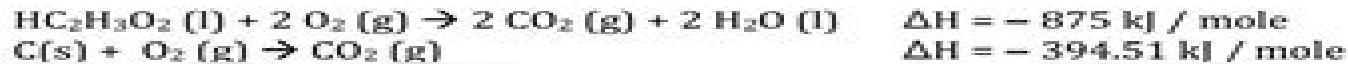
1. Calculate the ΔH for the reaction: $C_2H_4(g) + H_2(g) \rightarrow C_2H_6(g)$.
from the following data:



2. Calculate the ΔH for the reaction:



3. Find the ΔH_f° (heat of formation) for acetic acid, $HC_2H_3O_2$, using the following thermochemical data:



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**David Shephard Garland,James
Cockcroft,Lucius Polk
McGehee,Charles Porterfield**

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