

Chemistry Final Exam Review Questions

Useful Values and Equations

Some of the following values and equations may be useful for answering some of the questions and problems.

$$\text{op} = 1.84(\text{°C}) + 32$$

$$\text{speed of light} = 3.00 \times 10^8 \text{ m/s}$$

$$c = \lambda\nu$$

$$R = 0.082057 \text{ L-atm / deg-mole}$$

$$\text{Planck's constant} = 6.63 \times 10^{-34} \text{ J·s}$$

$$E = hv$$

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- One of the following does not describe gases. Which one is it?
 - Expands infinitely.
 - Rigid shape.
 - Easily compressed.
 - Relatively low densities.
- The following properties describe the element, mercury (Hg). Which one is a **chemical** property?
 - Its density is 5.43 g/cm³.
 - It is a silvery-white metal.
 - It is a liquid at 298 K.
 - Mercury(II) sulfide forms when it is exposed to sulfide ions.
- All of the following are properties of SO₂. Which one is a **physical** property?
 - It reacts with oxygen to form SO₃²⁻.
 - It condenses to a colorless liquid at 14°F and atmospheric pressure.
 - It is a strong reducing agent.
 - It reacts with water and oxygen to form an acid.
- Which of the following numbers has 4 significant figures?
 - 0.04309
 - 0.0430
 - 0.0431
 - 0.43900
- How many significant figures should there be in the answer to the following math equation?
 $2.834 + 5.71520 + 2.12 + 178.1 + 250.2619$
 - 3
 - 4
 - 5
 - 6
- How many significant figures should there be in the answer to the following math equation?
 $23.413 \text{ g} + (2.15 \text{ cm} \times 1.1 \text{ cm} \times 3.75 \text{ cm})$
 - 2
 - 3
 - 4
 - 5
- At 25°C, one cm³ of iron has a mass of 7.62 grams. What volume of iron is occupied by 12.5 grams of iron?
 - 0.82 cm³
 - 8.2 cm³
 - 0.164 cm³
 - 16.4 cm³
- A cube of lead has a volume of 6.61 cm³. Its mass is 75.0 grams. What is its density?
 - 3.85 g/cm³
 - 7.41 g/cm³
 - 11.3 g/cm³
 - 22.6 g/cm³

2nd Semester Chemistry Final Exam Review Answers

Daniel F McAuley

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Artificial Intelligence in Education. Posters and Late Breaking Results, Workshops and Tutorials, Industry and Innovation Tracks, Practitioners, Doctoral Consortium and Blue Sky Andrew M. Olney,Irene-Angelica

Chounta,Zitao Liu,Olga C. Santos,Ig Ibert Bittencourt,2024-07-01 This volume constitutes poster papers and late breaking results presented during the 25th International Conference on Artificial Intelligence in Education AIED 2024 which took place in Recife Brazil during July 8 12 2024 The 18 full papers and 92 short papers were carefully reviewed and selected from 200 submissions They are organized in topical sections as follows Part One Blue Sky Industry Innovation and Practitioner WideAIED and Late Breaking Results Part Two Late Breaking Results Doctoral Consortium Workshops and Tutorials

Nursing School Entrance Exam Review William H. Brown, Frederick A. Bettelheim, Jerry March,2004-04

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Prentice,2020-01-06 This journey will engage you in dealing with some hard truths and it will take you down a new pathway and new ways of thinking about K 12 education We now live in a nation that is struggling with deep social economic and political conflicts We are all doing our best to resolve these conflicts and to solve the critical challenges that we all face in the Digital Age but our children and young adults are having a very difficult time in dealing with the realities of their young lives We wrote this book because we want to engage all of our readers in each local community in frank honest down to earth practical conversations about our K 12 schools as the foundation for our constitutional democracy Without well educated citizens our government our economy and our society will not survive And this is true regardless of the political beliefs of our readers across the political spectrum

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Director's Handbook Arco Publishing Company, 1964 *The Chemistry Classroom* James Dudley Herron, 1996 Aimed at chemists who teach at the high school and introductory college level this valuable resource provides the reader with a wealth of knowledge and insight into Dr Herron's experiences in teaching and learning chemistry Using specific examples from chemistry to illustrate principles of learning the volume applies cognitive science to teaching chemistry and explores such topics as how individuals learn teaching problem solving concept learning language roles and task involvement Includes learning exercises to help educators decide how they should teach

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