

SECTION A

(Multiple Choice Questions)

- Answer all questions.
- For each question there are four possible answers, **A**, **B**, **C** and **D**.
- Choose the one you consider correct and mark your choice in **SOFT** pencil on the separate answer sheet.
- If you want to change an answer, thoroughly erase the one you wish to delete.
- On page 17 is a copy of the Periodic Table.

1. Which labelled arrow shows sublimation?



[1]

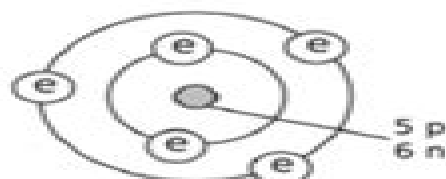
2. Which is an example of a compound?

- A** carbon
- B** diamond
- C** iron oxide
- D** mercury [1]

3. What are the correct charges for electrons, neutrons and protons?

	electron	neutron	proton
A	-1 0 +1		
B	+1 0 -1		
C	0 +1 -1		
D	+1 -1 0 [1]		

4. Which element's atomic structure is shown here?



- A** barium
- B** beryllium
- C** boron
- D** sodium [1]

Multiple Choice Review Questions Physical Science

**Jared Sparks,Edward Everett,James
Russell Lowell,Henry Cabot Lodge**

Multiple Choice Review Questions Physical Science:

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and Journal of Physical Science William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1919 [The Houghton Mifflin Guide to Reading Textbooks](#) Houghton Mifflin Co, 2004-05-05 The Houghton Mifflin Guide to Reading Textbooks highlights key skills and strategies required to successfully read college level materials Part One describes key elements that often appear in textbooks such as definitions visual aids and charts Part Two examines how to deal with distractions manage time take notes and read critically In Part Three students apply what they have learned to 5 short selections from various college disciplines Part Four features three full length textbook chapters from actual business physical sciences and history texts **A Student's Manual of a Laboratory Course in Physical Measurements** Wallace Clement Sabine, 1893

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, 1885 *Reviewing Physics* David R. Kiefer, 1995 *The Principles of Physics* Alfred Payson Gage, 1901 *Chemical News*

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