

**5.4****THE PERIODIC TABLE: ORGANIZING THE ELEMENTS**  
**SECTION REVIEW****Objectives**

- Describe the structure of the periodic table.
- Identify the position of groups, periods, and the transition metals in the periodic table.

**Key Terms**

■ periodic table	■ metal	■ nonmetal
■ group	■ metalloids	■ halogens
■ periodic law	■ noble gases	■ transition metals
■ period	■ lanthanides	■ actinides
■ halogen	■ lanthanide contraction	
■ noble gas	■ transition metal effects	
■ lanthanides	■ lanthanide contraction	
■ actinides	■ lanthanide contraction, lanthanides	

**Part A: Completion**

After their completion, complete the blank spaces after reading each concept and answer each question in the review. Each blank can be completed with one, two, or three words, or a name here.

The periodic table organizes elements into vertical \_\_\_\_\_ 1. \_\_\_\_\_ and horizontal \_\_\_\_\_ 2. \_\_\_\_\_ in order of increasing \_\_\_\_\_ 3. The table is made of vertical columns of elements. Those that are similar in chemical properties are in the same \_\_\_\_\_ 4. The elements in Group 1A (hydrogen) VA are called the \_\_\_\_\_ 5. The \_\_\_\_\_ 6. make up Group 0. The elements in Groups 13 and 15 are incorporated in periods 4 and 5 by the \_\_\_\_\_ 7. and in the periods 6 and 7 by the \_\_\_\_\_ 8. The Group 14 elements are called the \_\_\_\_\_ 9. and the Group 12 elements are called the \_\_\_\_\_ 10. The nonmetals of \_\_\_\_\_ 11. are called the halogens. Elements with properties that are intermediate between metals and nonmetals are called \_\_\_\_\_ 12. \_\_\_\_\_



## Answer Key To Organizing Elements

**Association for Quality and  
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