

NAME: _____

CLASS: _____

DATE: _____

1.



Terry fills two balloons with the same amount of helium gas. She places one of her helium balloons in the freezer. After twenty minutes, she removes the balloon from the freezer and notices that it is smaller than the balloon that stayed at room temperature. Her balloons are shown above.

How does Terry correctly explain the difference in the size her balloons?

- ☐ A She says that an increase in pressure increases the volume of helium gas.
- ☐ B She says that a decrease in pressure decreases the volume of helium gas.
- ☐ C She says that a decrease in temperature increases the volume of helium gas.
- ☐ D She says that an increase in temperature increase the volume of the helium gas.

2.



Using the periodic table, which of the following properties does Terry list for helium? Select all that apply.

- ☐ A Helium is a metalloid.
- ☐ B Helium has 2 valence electrons.
- ☐ C Helium is a metal.
- ☐ D Helium is a nonmetal.
- ☐ E Helium has 4 valence electrons.

3.



The chemical properties of calcium are MOST similar to the chemical properties of:

- ☐ A Sc
- ☐ B Mg
- ☐ C Ar
- ☐ D K

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**Michael J. Padilla, David V. Frank, John
G. Little, Steve Miller**



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