

Build An Atom PhET Lab SOLUTIONS

Introduction: Atoms are the smallest things that retain the properties of matter we can observe. Atoms are made of three subatomic particles; protons, neutrons, and electrons.

- Protons have a mass of _____ unit and a charge of _____.
- Neutrons have a mass of _____ unit and a charge of _____.
- Electrons have a mass of nearly _____ unit and a charge of _____.

In this simulation, you will build atoms, subatomic particle by subatomic particle and observe the effect of adding more of each particle. When the subatomic particles in an atom change, an ion, isotope or different element will be created.



Procedure: Play with the Sims: → Chemistry → Build An Atom [Background](#)

Begin by playing with the simulation for a while. Become familiar with the interface. What happens when you add protons, neutrons, or electrons? To start over, click **Reset All**.

Show the symbol, atomic mass, and charge by clicking on the

Analysis Questions

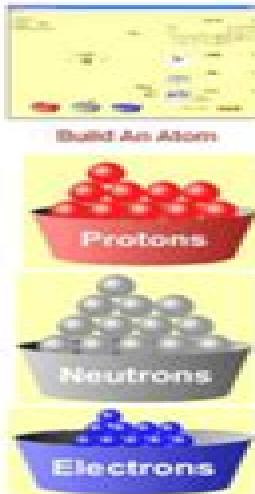
- Ions are atoms of the same element with different numbers of _____ electrons.
- Isotopes are atoms of the same element with different numbers of _____ neutrons.
- Adding or removing protons from an atom does what to the atom? _____ changes the atom or element's identity
- An atom with the same number of protons and electrons has a charge of _____.
- Adding two electrons to a neutral atom produces an ion with a charge of _____.
- An atom with six protons and five electrons would have a charge of _____.
- What atom is created with nine protons, nine neutrons, and nine electrons? **Fluorine**
- Show the full symbol for the above atom in the box at the right.
- What does the upper-left number in the symbol represent? **mass, protons and neutrons**
- What does the lower-left number in the symbol represent? **atomic #, protons**
- Draw the atoms described below, showing protons, neutrons, and electrons:

Hydrogen: H

Carbon: C

Oxygen: O

Neon: Ne



18 **F**
9

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Building An Atom Phet Simulation Sheet Answers

Franklin S. Allaire,Jennifer E. Killham

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