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Maintenance Manual

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web expert answer 100 2 ratings the example is based upon henderson hasselbalch equation the equation gives relation between ph and pka of weak acid the equation is $\text{pH} = \text{pK}_a + \log \frac{[\text{salt}]}{[\text{acid}]}$ at equivalence point all a view the full answer transcribed image text

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Volume of NaOH added (ml)	pH
4	6.4
6	4.6
5	5.9
5	5.9

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