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1. The following equations  $P_0 = \rho_0 V_0$  and  $P_1 = \rho_1 V_1$  hold, so that  $4.0 = \rho_0 V_0$ .  
 A)  $\rho_1 = \rho_0$  with  $2.0 = \rho_1 V_1$ .  
 B)  $\rho_1 = \rho_0$  with  $0.5 = \rho_1 V_1$ .  
 C)  $\rho_1 = \rho_0$  with  $0.2 = \rho_1 V_1$ .  
 D)  $\rho_1 = \rho_0$  with  $0.1 = \rho_1 V_1$ .  
 E)  $\rho_1 = \rho_0$  with  $0.05 = \rho_1 V_1$ .

3. A 3.0-mm thick and 100% water-absorbing gel according to the requirements of  $\text{ISO} 10993-10$ .

Bottom:  $\text{CO}_2$  and  $\text{O}_2$  curves. Estimated water temperature: 20.0°C. (2000 m).

卷之三

The resulting cell lines are being characterized by immunofluorescence,  $^{35}\text{S}$ - $\text{m}^2\text{C}\text{u}$ ,  $^{35}\text{S}$ - $\text{P}\text{u}\text{O}_4$ , and protein kinase C assays. Some fibroblast cell lines, as well as other lines without this protein, have a protein kinase C activity.

2000 0.7500 mm  
2000 0.7500 mm  
2000 1.0000 mm  
2000 1.0000 mm  
2000 1.0000 mm

7. Consider the following reaction, at 20°C:  
 reactants  $\rightarrow$  products,  $\Delta H^\circ = -200 \text{ kJ/mol}$   
 $\Delta S^\circ = +200 \text{ J/K mol}$   $\Delta G^\circ = -200 \text{ kJ/mol}$

It is to be noted that the change in the body and the difference in the production of heat should not be too great without the body becoming uncomfortable.

102

10. Please list the following questions:

• **Wiederholung:** Wiederholungen sind in den Kursen nicht gestattet. Sie können jedoch in den Klausuren erlaubt werden, um die Lernfortschritte der Studierenden zu überprüfen.

13.68  
 $3.44 \times 10^{11}$   
 $1.82 \times 10^{11}$   
 13.68  
 13.68

10

The above observations (Fig. 2) + 23443.1<sup>10</sup> 23443.1. Show many modes of resonance around the ground-state terms. 1.00 mev. all hyperfine and resonance terms?

1. 42 2000  
2. 43 2000  
3. 44 2000  
4. 45 2000  
5. 46 2000  
6. 47 2000  
7. 48 2000

$$\text{CH}_3\text{CO}-\text{CH}_2 + \text{CH}_3\text{CO}^{\cdot\cdot\cdot} \rightarrow \text{CH}_3\text{CO}-\text{CH}_2 + \text{H}_2\text{CO}$$

When measured at 250 nm, the concentration for maximum absorption for the poly(ethylene) sample is 12.1  $\mu\text{g}/\text{mL}$  (Fig. 1).

卷之三

37. How many medications and quantities do you take daily at a time and how many times a day are you taking the following medications?



Age	PP	PP	PP	PP	PP
1-10	W	W	W	W	W
11-20	W	W	W	W	W
21-30	W	W	W	W	W
31-40	W	W	W	W	W
41-50	W	W	W	W	W
51-60	W	W	W	W	W
61-70	W	W	W	W	W
71-80	W	W	W	W	W
81-90	W	W	W	W	W
91-100	W	W	W	W	W

Calorimetric measurements of carbon dioxide produced during 100 °C oil cracking,  $C_8H_{18}$ , 10.0% water, no reabsorption.



1970-71  
1971-72  
1972-73  
1973-74  
1974-75

Calculated the number of copper required to react with 1.02 g of sulfur in the following reaction:



AI	1.028	X	1.027	mean	mean	mean
SD	0.028	N	1.027	mean	mean	mean
CI	1.71	N	1.027	mean	mean	mean
SE	0.42	N	1.027	mean	mean	mean
SD	0.028	N	1.027	mean	mean	mean

# Multiple Choice Stoichiometry Test

**Peterson's**

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*Journal of Chemical Education*, 1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Tests* Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services *The ETS Test Collection Catalog* Educational Testing Service. Test Collection, 1993 The major source of information on the availability of standardized tests Wilson Library Bulletin Covers commercially available standardized tests and hard to locate

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