

THE KINETIC ENERGY/MOLECULE TEST (Version A)

Given the following record back to complete the sentences.

Molecular Data						
Identified,	collide,	Temperature,	in molecules,	Position,	equilibrium,	Total Kinetic energy
1. Things are made of _____.				(or other types of atoms/groups)		
2. When a thing gets _____, its molecules are moving faster and have more kinetic energy.				(or molecules are moving slower and have less kinetic energy)		
3. When a thing gets _____, its molecules are moving slower and have less kinetic energy.				(or molecules are moving faster and have more kinetic energy)		
4. _____ is a measure of the average kinetic energy of the molecules of a thing.						
5. Energy isn't created or _____.						
6. Energy can't decrease in a system as it decreases in another part of the system. If the total energy of a system doesn't change,						
7. The molecules of a system will spread energy until the system reaches a stable state known as _____, in which all of the molecules are moving all about the system again.						
8. If things at the same temperature, the thing with more molecules has more _____ (kinetic energy) than the thing with fewer molecules.						

Complete the following multiple choice questions by clicking the best answer.

1. Lucy cooks a pot of soup, then puts the soup in the freezer for a while. Now the soup in the pot is hot, but the soup in the freezer is cold. What is the difference between the molecules of the soup in the pot and the molecules of the soup in the freezer?
 - The molecules of the soup in the pot are moving and the molecules of the soup in the freezer are not moving.
 - The molecules of the soup in the pot are moving faster than the molecules of the soup in the freezer.
 - There is no difference because all of the soup is made of the same type of molecules.
 - The molecules of the soup in the freezer are smaller than the molecules of the soup in the pot.
2. Engineers design computers to cool them down quickly so they don't break. What happens to the molecules in a computer when the temperature of the computer decreases?
 - The energy of the molecules in the computer decreases.
 - The energy of the molecules in the computer increases.
 - The cold energy of the molecules in the computer increases.
3. Joseph puts a piece of bread on a plate. The bread is hot from the oven. When he puts the bread on the plate, the energy of the plate increases. What must be true of the energy of the bread?
 - The energy stays the same while the energy of the plate increases.
 - The energy decreases while the energy of the plate increases.
 - The energy increases while the energy of the plate decreases.

Blue Battery Test Study Guide

**Charles Alpheus Bennett, William
Thomas Bawden**

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Vocational Division Bulletin ,1961 *Vocational Education Bulletin* United States. Division of Vocational Education,1961
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Amanda Wingfield is a faded, tragic remnant of Southern gentility who lives in poverty in a dingy St. Louis apartment with her son, Tom, and her daughter, ... The Glass Menagerie When Amanda convinces Tom to bring home from his workplace a "gentleman caller" for Laura, the illusions that Tom, Amanda, and Laura have each created in order ... The Glass Menagerie Text Scene 1: The Wingfield apartment is in the rear of the building, one of those vast hive-like conglomerations of cellular living-units that flower as. Tennessee Williams - The Glass Menagerie (Scene 3) LEGEND ON SCREEN: 'AFTER THE FIASCO' [TOM speaks from the fire-escape landing.] TOM: After the fiasco at Rubicam's Business College, the idea of getting a ... "The Glass Menagerie," Scene One and Scene Two, by ... 41 Scene 1. 352 The Wingfield apartment is in the rear of the building, one of those vast hive-like conglomerations of cellular living-units that flower as ... Tennessee Williams - The Glass Menagerie (Scene 7) A moment after the curtain rises, the lights in both rooms flicker and go out.] JIM: Hey, there, Mr Light Bulb ! [AMANDA laughs nervously. LEGEND: 'SUSPENSION ... The Glass Menagerie: Acting Edition: Tennessee Williams A new introduction by the editor of The Tennessee Williams Annual Review, Robert Bray, reappraises the play more than half a century after it won the New York ...