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STOICHIOMETRY PRACTICE WORKSHEET

1. Using the following equation:



Calculate how many grams of iodine are needed to prepare 28.6 grams of ICl by this reaction.

2. Using the following equation:



How many moles and grams of KMnO₄ are needed for this reaction on 11.4 grams of KNO₂?

3. Using the following equation:



How many moles and grams of oxygen (O₂) are needed to react with 56.8 grams of ammonia by this reaction?

4. Using the following equation:



Calculate the number of moles and the number of grams of iodine (I₂) that can be made this way from 16.4 grams of NaIO₃.

Basic Stoichiometry Practice Problems

Charles H. Corwin

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The Magic of Molecules:

Exploring Chemical Reactions Dr. M. Ziauddin Shahzada ,2025-08-25

Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry Malone,2000-04-18

Improving Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner,2007

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