

**CHAPTER 7****STUDY GUIDE****Ionic Compounds and Metals****Section 7.1 Ion Formation**

In your textbook, read about chemical bonds and formation of ions.

Use each of the terms below just once to complete the passage.

chemical bond	electrons	energy level	ions	noble gases
nucleus	octet	pseudo-noble gas formations	valence	

The force that holds two atoms together is called a(n) (1) \_\_\_\_\_.

Such an attachment may form by the attraction of the positively charged

(2) \_\_\_\_\_ of one atom for the negatively charged

(3) \_\_\_\_\_ of another atom, or by the attraction of charged atoms, which are called

(4) \_\_\_\_\_. The attractions may also involve

(5) \_\_\_\_\_ electrons, which are the electrons in the outermost

(6) \_\_\_\_\_. The (7) \_\_\_\_\_ are a family of elements that have very

little tendency to react. Most of these elements have a set of eight outermost electrons, which is called a

stable (8) \_\_\_\_\_. The relatively stable electron structures developed by loss of

electrons in certain elements of groups 3, 4, 13, and 14 are called (9) \_\_\_\_\_.

For each statement below, write *true* or *false*.

\_\_\_\_\_ 10. A positively charged ion is called an anion.

\_\_\_\_\_ 11. Elements in group 1 lose their one valence electron, forming an ion with a 1+ charge.

\_\_\_\_\_ 12. Elements tend to react so that they acquire the electron structure of a halogen.

\_\_\_\_\_ 13. A sodium atom tends to lose one electron when it reacts.

\_\_\_\_\_ 14. The electron structure of a zinc ion ( $Zn^{2+}$ ) is an example of a pseudo-noble gas formation.

\_\_\_\_\_ 15. A  $Cl^-$  ion is an example of a cation.

\_\_\_\_\_ 16. The ending *-ide* is used to designate an anion.

\_\_\_\_\_ 17. Nonmetals form a stable outer electron configuration by losing electrons and becoming anions.

# Answers To Ionic Compound Study Guide

**Susan McMurry, John McMurry**

## Answers To Ionic Compound Study Guide:

*Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

*Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today* Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application.

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd numbered exercises.

*Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition* John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter.

**Study Guide and Solutions Manual to Accompany Organic Chemistry, 11th Edition** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2013-03-25 This is the study guide and solutions manual to accompany Organic Chemistry 11th Edition.

*Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry* Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana. Sue Baker Publishing News

**Study Guide with Answers to Selected Problems** H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test. The answers section includes answers for all odd numbered end of chapter exercises.

**Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Denniston Katherine, 2006

*Study Guide and Full Solutions Manual*

Susan McMurry, John McMurry, 2002-08 Contains a brief overview of every chapter review of skills

self tests and the answers and detailed solutions to all end of chapter problems in the textbook **Study Guide/Selected Solutions Manual** Julie R. Frentrup,2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book **Study Guide with Selected Solutions** Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems **Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e** Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Study Guide & Solutions Manual to Accompany Organic Chemistry** Mary H. Bailey,1998 Study Guide & Solutions Manual to Accompany Organic Chemistry, Third Edition G. Marc Loudon,Joseph G. Stowell,1995 **Study Guide General Chemistry, Second Edition, Becker/Wentworth** Becker,Benjamin L. Carroll,1980

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