

TAP Chemistry Fall Semester Review Sheet**Lab Safety & Equipment:**

1. What should you do if you get acid on your hands? **Rinse with lots of water**
2. How should you smell a chemical if you are instructed to do so? **waft**
3. When are goggles required? **When using chemicals, heat or glassware**
4. What should you do with broken glass? **Dispose of broken glass in the broken glass container**

Kinetic Theory & States of Matter:

5. What are the three main states of matter? **Solids, liquids & gases**
6. Fill out the table below for the three states of matter

State	Compressibility (high/low)	Shape (definite/indefinite)	Volume (definite/indefinite)	Kinetic energy (high/low)
Solid	LOW	DEFINITE	DEFINITE	LOW
Liquid	LOW	INDEFINITE	DEFINITE	MEDIUM
Gas	HIGH	INDEFINITE	INDEFINITE	HIGH

7. Which phase is the most structured? The least? **Solids are the most structured, Gases are the least**
8. Match the following phase changes:

<u>C</u> Melting <u>D</u> Freezing <u>A</u> Boiling/Evaporation <u>F</u> Condensation <u>B</u> Sublimation <u>E</u> Deposition	A. change in phase from a liquid to a gas B. change in phase from a solid to a gas C. change in phase from a solid to a liquid D. change in phase from a liquid to a solid E. change in phase from a gas to a solid F. change in phase from a gas to a liquid
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Classification of Matter:

9. Identify each of the following as either an

Element, Compound, Homogeneous Mixture or Heterogeneous Mixture

- | | |
|---|--|
| A. Pizza _____ Heterogeneous _____ | C. Salt (NaCl) _____ Compound _____ |
| B. Vanilla Ice Cream _____ Homogeneous _____ | D. Carbon _____ Element _____ |

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**Queensland. Parliament. Legislative
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