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Chemistry Sem 1 Apex Learning

6.1.2 Exam: Semester 1 Exam

Question 6 of 25

What is the net ionic equation for the reaction between Na_2SO_4 and CaCl_2 ?

- A. $\text{Na}_2\text{SO}_4(\text{aq}) + \text{CaCl}_2(\text{aq}) \rightarrow \text{CaSO}_4(\text{s}) + 2\text{NaCl}(\text{aq})$
- B. $2\text{Na}^+(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow 2\text{NaCl}(\text{aq})$
- C. $\text{SO}_4^{2-}(\text{aq}) + \text{Ca}^{2+}(\text{aq}) \rightarrow \text{CaSO}_4(\text{s})$
- D. $2\text{Na}^+(\text{aq}) + \text{SO}_4^{2-}(\text{aq}) + \text{Ca}^{2+}(\text{aq}) + 2\text{Cl}^-(\text{aq}) \rightarrow \text{CaSO}_4(\text{s}) + 2\text{Na}^+ + 2\text{Cl}^-(\text{aq})$

Answer

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M Carnoy

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