

## Chapter 2 Active Reading Guide

### The Chemical Context of Life

This chapter covers the basics that you may have learned in your chemistry class. Whether your teacher goes over this chapter, or assigns it for you to review on your own, the questions that follow should help you focus on the most important points.

#### Section 1

1. Define and give an example of the following terms:

Term	Definition
matter:	All physical substances.
element:	Chemical substances that cannot be broken down
compound:	Mixture of two or more elements.

2. What four elements make up 96% of all living matter?

Carbon, Oxygen, Hydrogen, and Nitrogen

3. What is the difference between an *essential element* and a *trace element*?

While essential elements are needed in large quantities for an organism to live a healthy life, trace elements are needed in very little amounts.

#### Section 2

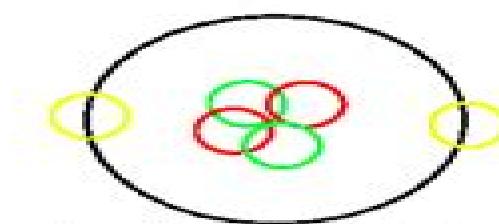
4. Sketch a model of an atom of helium, showing the electrons, protons, neutrons, and atomic nucleus.



=Proton



Neutron



5. What is the atomic number of helium? \_\_\_\_\_ Its atomic mass? \_\_\_\_\_

# Ap Biology Reading Guide Answers Chapter 25

**Stephen J. McPhee**

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Sylvia S. Mader,2003-08 This best selling text emphasizes the relationship between humans and other living things Intended for an introductory course this text provides students with a firm grasp of how their bodies function and how the human population can become more fully integrated into the biosphere An Online Learning Center tied directly to the text via icons will direct students to activities or animations that gives a visual example of difficult processes as well as Working Together boxes to emphasize homeostasis **Handbook of Animal Models in Neurological Disorders** Colin R. Martin,Vinood B.

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