

Acids and bases review sheet Answer key

1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Acid	base
Conducts electricity	conducts electricity
Contains hydrogen ion	contains hydroxide ion
Proton donor	proton acceptor
Turns litmus paper red	turns litmus paper blue
React with most metals	releases hydroxide ion
Releases hydronium	Ex: Windex
Ex: lemon	bleach
Vinegar	soap
Battery acid (sulfuric acid)	

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, acids or bases:

- a)  A produce hydrogen ions ( $H^+$ ) in solution
- b)  B have a bitter taste
- c)  B produce hydroxide ions ( $OH^-$ ) ions
- d)  A react with bases to form salts
- e)  A have a sour taste
- f)  B have a slippery, soapy feel
- g)  B react with acids to form salts
- h)  A react with many metals

3. Write an equation for the dissociation of the following acids in water. Follow example.

a)  $\text{HClO}_4 \rightarrow \text{H}^+ + \text{ClO}_4^-$       d)  $\text{H}_2\text{S} \rightarrow 2\text{H}^+ + \text{S}^{2-}$   
 b)  $\text{H}_2\text{SO}_4 \rightarrow 2\text{H}^+ + \text{SO}_4^{2-}$       e)  $\text{HCl} \rightarrow \text{H}^+ + \text{Cl}^-$   
 c)  $\text{HC}_2\text{H}_5\text{O}_2 \rightarrow \text{H}^+ + \text{C}_2\text{H}_5\text{O}_2^-$       f)  $\text{HNO}_3 \rightarrow \text{H}^+ + \text{NO}_3^-$

4. Write an equation for the dissociation (ionization) of the following bases in water: (follow example).

a)  $Mg(OH)_2 \rightarrow Mg^{2+} + 2OH^-$   
b)  $NaOH \rightarrow Na^+ + OH^-$   
c)  $Ba(OH)_2 \rightarrow Ba^{2+} + 2(OH^-)$   
d)  $KOH \rightarrow K^+ + OH^-$   
e)  $Mg(OH)_2 \rightarrow Mg^{2+} + 2(OH^-)$   
f)  $LiOH \rightarrow Li^+ + OH^-$

Q: What type of reaction is a neutralization reaction? Double replacement

B. Write the net equation for neutralization:



# Acid Base Review Answers

**Linda Anne Silvestri**

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### **Table of Contents Acid Base Review Answers**

1. Understanding the eBook Acid Base Review Answers
  - The Rise of Digital Reading Acid Base Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Acid Base Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Acid Base Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Acid Base Review Answers
  - Personalized Recommendations

- Acid Base Review Answers User Reviews and Ratings
- Acid Base Review Answers and Bestseller Lists

5. Accessing Acid Base Review Answers Free and Paid eBooks

- Acid Base Review Answers Public Domain eBooks
- Acid Base Review Answers eBook Subscription Services
- Acid Base Review Answers Budget-Friendly Options

6. Navigating Acid Base Review Answers eBook Formats

- ePUB, PDF, MOBI, and More
- Acid Base Review Answers Compatibility with Devices
- Acid Base Review Answers Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Acid Base Review Answers
- Highlighting and Note-Taking Acid Base Review Answers
- Interactive Elements Acid Base Review Answers

8. Staying Engaged with Acid Base Review Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Acid Base Review Answers

9. Balancing eBooks and Physical Books Acid Base Review Answers

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Acid Base Review Answers

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Acid Base Review Answers

- Setting Reading Goals Acid Base Review Answers
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Acid Base Review Answers

- Fact-Checking eBook Content of Acid Base Review Answers

- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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