

I. Answer the following:

1. Beta particles is designated as ${}_{-1}^0e$
2. Half-life of an atom is a period of decay of one half of the initial atoms
3. α particles have the least penetration power.
4. ${}_{92}^{238}\text{U} \rightarrow {}_2^4\text{He} + {}_{90}^{234}\text{Th}$ is a/an alpha decay.
5. The relationship between decay constant and half of radioactive nuclei is $T_{1/2} = 0.693/\lambda$.

II. Answer in short:

1. Radioactivity: It is a property exhibited by certain types of matter of emitting energy and subatomic particles simultaneously. Radioactivity can be seen in such forms:
 - alpha decay,
 - beta decay,
 - gamma decay.

2. Nuclear fission and nuclear fusion.

<u>Nuclear fission</u>	<u>Nuclear fusion</u>
A nuclear fission reaction is splitting up of a large atom or a molecule into two or more smaller ones.	A nuclear fusion is the process of process of combination of two or more lighter atoms or molecules into larger ones.
* <u>Example</u> : Atomic bomb	* <u>Example</u> : Hydrogen bomb

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1. Understanding the eBook Advanced Inorganic Chemistry Study Guide
 - The Rise of Digital Reading Advanced Inorganic Chemistry Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Advanced Inorganic Chemistry Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Advanced Inorganic Chemistry Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Advanced Inorganic Chemistry Study Guide
 - Personalized Recommendations
 - Advanced Inorganic Chemistry Study Guide User Reviews and Ratings
 - Advanced Inorganic Chemistry Study Guide and Bestseller Lists

5. Accessing Advanced Inorganic Chemistry Study Guide Free and Paid eBooks
 - Advanced Inorganic Chemistry Study Guide Public Domain eBooks
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 - Advanced Inorganic Chemistry Study Guide Budget-Friendly Options
6. Navigating Advanced Inorganic Chemistry Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Advanced Inorganic Chemistry Study Guide Compatibility with Devices
 - Advanced Inorganic Chemistry Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Advanced Inorganic Chemistry Study Guide
 - Highlighting and Note-Taking Advanced Inorganic Chemistry Study Guide
 - Interactive Elements Advanced Inorganic Chemistry Study Guide
8. Staying Engaged with Advanced Inorganic Chemistry Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Advanced Inorganic Chemistry Study Guide
9. Balancing eBooks and Physical Books Advanced Inorganic Chemistry Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Advanced Inorganic Chemistry Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Advanced Inorganic Chemistry Study Guide
 - Setting Reading Goals Advanced Inorganic Chemistry Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Advanced Inorganic Chemistry Study Guide
 - Fact-Checking eBook Content of Advanced Inorganic Chemistry Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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