

Acids, Bases and Salts

- **THE THEORY of ACIDS and ALKALIS** and a few technical terms:
 - **Acids** are substances that **form hydrogen ions ($\text{H}^+_{(\text{aq})}$)** when dissolved in water eg hydrochloric acid HCl gives $\text{H}^+_{(\text{aq})}$ and $\text{Cl}^-_{(\text{aq})}$ ions, sulphuric acid H_2SO_4 gives $2\text{H}^+_{(\text{aq})}$ and SO_4^{2-} ions and nitric acid HNO_3 gives $\text{H}^+_{(\text{aq})}$ and $\text{NO}_3^-_{(\text{aq})}$ ions.
 - **Alkalis** are substances that **form hydroxide ions ($\text{OH}^-_{(\text{aq})}$)** in water eg sodium hydroxide NaOH gives $\text{Na}^+_{(\text{aq})}$ and $\text{OH}^-_{(\text{aq})}$ ions, calcium hydroxide $\text{Ca}(\text{OH})_2$ gives $\text{Ca}^{2+}_{(\text{aq})}$ and $2\text{OH}^-_{(\text{aq})}$ ions. Note: **an alkali is a base soluble in water.**
 - In **water**, there are trace quantities of H^+ and OH^- ions BUT they are of equal concentration and so water is neutral.
 - In **acid solutions** there are **more H^+ ions than OH^- ions.**
 - In **alkaline solution** there are **more OH^- ions than H^+ ions.**
 - Acids dissociate to different extents in aqueous solution. Acids that dissociate to a large extent are strong electrolytes and **strong acids**. In contrast, acids that dissociate only to a small extent are **weak acids** and weak electrolytes.

In a similar manner, bases can be strong or weak depending on the extent to which they dissociate and produce OH^- ions in solution. Most metal hydroxides are strong electrolytes and **strong bases**. Ammonia, NH_3 , is a weak electrolyte and **weak base**.



- **BASES** eg oxides and hydroxides are substances that **react and neutralise acids to form salts and water**. **Bases which are soluble in water are called alkalis.**

Acids

Some common acids are listed below:

Name	Formula	Strong/Weak	Where is it found?
Hydrochloric acid	HCl	Strong	The stomach, in the lab.
Sulphuric acid	H_2SO_4	Strong	Acid rain, car batteries, the lab.
Nitric acid	HNO_3	Strong	Acid rain, in the lab.
Ethanoic (acetic) acid	CH_3COOH	Weak	Vinegar
Methanoic (formic) acid	HCOOH	Weak	Ant & nettle stings, descalers
Citric Acid	$\text{C}_6\text{H}_8\text{O}_7$	Weak	Citrus fruits

- Acids taste sour (e.g. vinegar, lemon juice).
- Acids are harmful to living cells.
- Aqueous solutions of all acids contain **hydrogen ions, H^+** .

Acids Bases And Salts Study Guide

Steven S. Zumdahl, Paul B. Kelter



Acids Bases And Salts Study Guide:

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Study Guide with Answers to Selected Problems

H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

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Henold,1984 **A Study Guide to Chemical Principles** Wilbert Hutton,1970 *Study Guide for General Chemistry for Colleges* Herman Thompson Briscoe,1936 Fundamentals of Medicinal Chemistry and Drug Metabolism M. O. Faruk Khan,Ashok Philip,2018-06-01 The primary objective of this 4 volume book series is to educate PharmD students on the subject of medicinal chemistry The book set serves as a reference guide to pharmacists on aspects of chemical basis of drug action This first volume of the series is comprised of 8 chapters focusing on basic background information about medicinal chemistry It takes a succinct and conceptual approach to introducing important fundamental concepts required for a clear understanding of various facets of pharmacotherapeutic agents drug metabolism and important biosynthetic pathways that are relevant to drug action Notable topics covered in this first volume include the scope and importance of medicinal chemistry in pharmacy education a comprehensive discussion of the organic functional groups present in drugs and information about four major types of biomolecules proteins carbohydrates lipids nucleic acids and key heterocyclic ring systems The concepts of acid base chemistry and salt formation and their applications to the drug action and design follow thereafter These include concepts of solubility and lipid water partition coefficient LWPC isosterism stereochemical properties mechanisms of drug action drug receptor interactions critical for pharmacological responses of drugs and much more Students and teachers will be able to integrate the knowledge presented in the book and apply medicinal chemistry concepts to understand the pharmacodynamics and pharmacokinetics of therapeutic agents in the body

The Top Books of the Year Acids Bases And Salts Study Guide The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the engaging narratives that have captivated audiences this year. Acids Bases And Salts Study Guide : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Acids Bases And Salts Study Guide : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Pape, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and thrilling novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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