

Acid Base test review Sheet

1. List three properties of acids: _____
2. List three properties of bases: _____
3. Acids turn litmus _____ and bases turn it _____
4. Name the following acids: Put a check next to it if it is a strong acid
 HBr _____ H₂SO₄ _____ HClO₄ _____
 HNO₃ _____ HI _____ H₂S _____
 HNO₂ _____
6. What is the formula for carbonic acid? _____ chloric acid _____ hydrochloric acid _____
8. What is the formula for finding the pH of an acid? _____
9. What is the [H₃O⁺] of a neutral solution? _____ what is the [OH⁻] of a neutral solution? _____
10. What is the K_w of water? _____
 Define :
 11. Arrhenius Acid _____
 12. Arrhenius base _____
 13. Bronsted-Lowry acid _____
 14. Bronsted-Lowry base _____
 15. Lewis acid _____
 16. Lewis base _____
 17. indicator _____
 18. titration _____
 19. Fill in the following table:

[H ⁺]	pH	[OH ⁻]	pOH	Acid or base
10 ⁻¹¹				
	5.6			
		2.27 x 10 ⁻⁵		
			8.33	

20. The strongest acid would have a pH of _____, the strongest base would have a pH of _____
21. What are the products of a neutralization reaction?
22. Write the balanced equation for the neutralization of Ca(OH)₂ and HCl

Acid And Base Test Review Sheet Answers

Joan I. Senyk, Darrell D. Ebbing

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